

Communication

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Concise Synthesis of Spirocyclic, Bridged γ -Butyrolactones via Stereospecific, Dyotropic Rearrangements of β -Lactones Involving 1,2-Acyl and δ -Lactone Migrations

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The simultaneous or stepwise reorganization of functional groups on vicinal carbon atoms via a 1,2-shift constitutes a type I dyotropic rearrangement, a term first proposed by Reetz in 1972.¹ One of the earliest examples of this rearrangement was the mutarotation of vicinal dibromides in steroidal systems first studied by Winstein and Barton.² Seminal work by Mulzer employing Lewis acid mediated versions with β -lactone substrates provided insights into the relative migratory aptitude of electron-rich groups in these type I dyotropic rearrangements.³ A requirement for such a molecular rearrangement is an antiperiplanar relationship between the migrating group and the β -C-O σ -bond of the β -lactone. Subsequent studies by Black⁴ and Reetz⁵ provided further information on the scope of this process for acyclic systems; however, debate remains regarding the degree of concertedness of the process.⁶ We recently reported a biscyclization of ketoacids that provides access to bicyclic and tricyclic β -lactones.⁷ Herein, we report the first examples of dyotropic rearrangements involving 1,2migrations of electron-deficient groups leading to spirocyclic, bridged γ -lactones 3 from fused tricyclic β -lactones 2, available from diketoacids 1 via biscyclizations (Figure 1). The first asymmetric variant of the latter process is also described.

We began our studies with tricyclic β -lactone **2a** (see Table 1),⁸ available from biscyclization of a cyclohexanedione precursor by our previously described procedure.7,9 We surmised that Lewis acid activation of the β -lactone would promote a dyotropic rearrangement via 1,2-acyl migration, leading to simultaneous, stereospecific¹⁰ ring expansion of the β -lactone to a γ -lactone with concomitant ring contraction of the cyclohexanone. The sole diastereomer isolated from the biscyclization, β -lactone 2a, possesses the required antiperiplanar relationship between the migrating C_4-O_1 bond of the β -lactone and the C_5 -C=O bond (relative stereochemistry confirmed by X-ray analysis⁹). Use of commonly employed conditions involving MgBr₂•Et₂O in Et₂O gave only trace conversion; however, the use of CH₂Cl₂ improved conversion to the expected spiro- γ -lactone (±)-3a after 18 h (Table 1, entry 1). Other Lewis acids (e.g., AlBr₃, Et₂AlCl, LaCl₃, PrCl₃, YCl₃) resulted in either complex mixtures or no reaction. Employing Yb(OTf)₃ and In(OTf)₃ suggested the beneficial effects of triflate over halide ligands (entries 2 and 3). In addition, a major improvement in conversion was realized with Zn(II) salts (entries 4 and 5) and 1.1 equiv of Zn(OTf)₂ provided **3a** in 94% yield (entry 6). Substoichiometric amounts of Zn(OTf)₂ were ineffective in promoting catalytic turnover on practical time scales (entry 7). Surprisingly, Mg(OTf)₂ did not promote this process (entry 8). Inspired by recent studies of Fuchs,^{6a} subjecting β -lactone **2a** to catalytic TMSOTf provided only ring-opened diacid 4 (73%, entry 9) suggestive of an intervening Grob-type fragmentation.¹¹

Additonal tricyclic β -lactones **2b**-**d**, available in varying yields and diastereoselectivities via biscyclization of dione acids **1b**-**d**,⁹ were also studied, and the major diastereomers gave excellent yields of bridged γ -lactones **3b**-**d** (Table 2, entries 2–4) with high stereospecificity (dr > 19:1).



Figure 1. Sequential biscyclization/dyotropic rearrangement route to tricyclic spiro- γ -lactones.

Table 1. Survey of Lewis Acids for Dyotropic Rearrangement of Tricyclic β -Lactone **2a**

	Me (+)-2a	Lewis acid CH ₂ Cl ₂ , 23 °C	(+)-3a	HO ₂ C C () Me 4	O₂F ∕
entry	Lewis acid ^a	equiv	<i>T</i> (°C)	time (h)	% yield (±)-3a ^b
1	MgBr ₂ •Et ₂ O	1.5	0→23	18	$ND^{c} (2:1)^{d}$
2	Yb(OTf) ₃	1.1	23	12	ND $(3:2)^d$
3	In(OTf) ₃	1.1	23	12	ND $(1:9)^d$
4	$ZnCl_2^e$	1.1	23	1.5	87
5	ZnBr ₂	1.1	23	1.5	90
6	Zn(OTf) ₂	1.1	0→23	1.5	94
7	$Zn(OTf)_2$	0.5	0→23	12	ND $(1:1)^d$
8	$Mg(OTf)_2$	1.1	23	12	NR^{f}
9	TMSOTf	0.2	23	12	$(73)^{g}$

^{*a*} All reactions were conducted at 0.05 M. ^{*b*} Refers to isolated yields. ^{*c*} ND = not determined. ^{*d*} Ratios in parentheses are for **3a/2a** as determined by ¹H NMR (500 MHz) analysis of crude reaction mixtures. ^{*e*} 1.0 M solution in CH₂Cl₂. ^{*f*} NR = no reaction. ^{*g*} Yield for diacid **4** accompanied by trace amounts of **3a**.

A plausible synchronous mechanism invokes a two-electron threecentered acylium intermediate 6,³ reminiscent of transition states proposed for Friedel–Crafts acylation,¹² or a carboxylate-stabilized, four-membered transition state 7, both benefiting from a homoconjugated, carbonyl-assisted migration (frangomeric effect) (Figure 2).¹³ To the best of our knowledge, 1,2-acyl migrations are unprecedented in type I dyotropic processes. Formation of diacid 4 with TMSOTF (Table 1, entry 9) points to a mechanistic extreme (i.e. 9) involving a stepwise dyotropic rearrangement leading to Grob-type fragmentation¹¹ that currently cannot be excluded as a pathway to **3a** when employing Zn(II) Lewis acids.¹⁴

Toward optically active tricyclic γ -lactones, we prepared β -lactone (–)-**2a** (53%, 97% ee, chiral GC⁹) via nucleophile-promoted desymmetrization of diketone **1a** with stoichiometric tetramisole hydrochloride (Scheme 1), recently employed by Birman for enantioselective acylations.¹⁵ This provides the first direct evidence for nucleophile involvement in the stereochemical setting step of biscyclizations with ketoacid substrates.^{7,16} Dyotropic rearrangement then gave γ -lactone (–)-**3a** in excellent yield with high stereochemical fidelity (99% ee, chiral GC).

Finally, we envisioned a $1,2-\delta$ -lactone dyotropic rearrangement with application toward natural product synthesis (Scheme 2). Despite the





^a Relative stereochemistry of 2a, 2b, 2c, 2c', 3a, and 3c was verified by X-ray analysis (ref 9). ^b Refers to isolated yields. ^c Ratios determined by ¹H NMR (500 MHz) of crude reaction mixtures. ^d Accompanied by 3% elimination byproduct (see ref 17). ^e Using major diasteromers 2c and 2d as substrates



Figure 2. Proposed mechanistic pathways for dyotropic 1,2-acyl shift and Grob-type fragmentation of β -lactone 2a.





potential reactivity of the β -lactone nucleus, we attempted a Baeyer–Villiger oxidation of β -lactone 2e using buffered conditions.^{6a} Although requiring long reaction times, oxidation proceeded smoothly to give the desired bislactone 12 in 74% yield (X-ray analysis, inset Scheme 2). While conditions described above gave only recovered 12, a brief survey of Lewis acids led to use of substoichiometric TMSOTf, which slowly produced the desired spiro- γ -lactone 13 constituting the core of curcumalactone (X-ray analysis, inset Scheme 2)

In summary, the repertoire of dyotropic rearrangements with β -lactone substrates has been expanded to include stereospecific 1,2shifts of both acyl and δ -lactone groups. This process, in conjunction with the enantioselective biscyclization of ketoacids, enables rapid Scheme 2. Synthesis and Dyotropic Rearrangement of Bislactone 12 (X-ray Structures of 12 and 13 Are Shown) Providing Spiro[5.5]lactone 13



construction of molecular complexity from simple ketoacid substrates in the form of novel spirocyclic, bridged keto- γ -lactones bearing a quaternary carbon adjacent to a tertiary alcohol center.

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Supporting Information Available: Experimental procedures and characterization data for tricyclic β -lactones **2a**–**d**, spiro- γ -lactones **3a**–**d**, bislactones 12, 13, including ¹H and ¹³C NMR spectra, chiral GC traces of (-)-2a and (-)-3a, and single-crystal X-ray structures. This material is available free of charge via the Internet at http://pubs.acs.org.

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